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American Cyanamid Company One Cyanamid Plaza Wayne, NJ 07470

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H. Michael D. Utidjian, M.D. Corporate Medical Director

November 3, 1993

INIT 11/04/93

Document Processing Center/(TS-790 ATTN: SECTION 8(E) COORDINATOR U.S. Environmental Protection Agend 401 M Street, S.W. Washington, D.C. 20460

Dear Sir/Madam:

The purpose of this letter is to inform you under Section 8(e) of TSCA of the study "Daphnia, Acute Immobilization Test" on a commercial cationic polymer mixture. The mixture has the following composition:

CAS#	Chemical Name	9
007732-18-5	Water	-° ~50
042751-79-1	Dimethylamine-Epichlorhydrin-	
	Ethylenediamine Polymer	~10

This study reports a 48-Hour Static EC50 of 0.6 mg/l with a no-effect level at 96 hours of 0.2 mg/ĭ.

EC50 determinations without suspended solids overestimates the true toxicity of cationic polymers. Suspended solids and other dissolved organic materials like humic acid which are present in natural waters reduce the effective concentration of the polymer and thereby its toxicity.

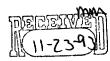
It is our understanding that the EPA is aware of the "mechanical" nature of the toxicity produced by cationic polymers and therefore, this information confirms data already known to the agency.

A final report of this study is enclosed. This document does not contain confidential business information.

Please direct all communications on this subject to Patricia Ann Vernon, Associate Toxicologist at the above address or call her at (201) 357-3375.

Sincerely,

H. M. Widyam, XH.
H. Michael D. Utidjian, M.D. Corporate Medical Director



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PROJECT NUMBER: 144442

DAPHNIA ACUTE IMMOBILIZATION TEST

TEST MATERIAL: MRD-92-444 (CT-519-92O)

PERFORMED AT:

EXXON BIOMEDICAL SCIENCES, INC. ENVIRONMENTAL TOXICOLOGY LABORATORY METTLERS RD. CN 2350 EAST MILLSTONE, NEW JERSEY 08875-2350

COMPLETION DATE: OCTOBER 13, 1993

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APPROVAL SIGNATURES

M. E. Jargia	_13 Oct 23
M. E. Targia, B.A. Study Director	DATI
Environmental Toxicology Laboratory	
D. H. Wasserstrom M.E.	8 OCT 1993
Director of Environmental Toxicology	DATE
L. D. Twitty, A.S.	12/0ct/93
L. D. Twitty, A.S. Analytical and Fate Chemistry Supervisor	DATE

I hereby declare to the best of my knowledge, this study was conducted in accordance with the OECD Principles of Good Laboratory Practice set forth in C(81)30 (Final), Annex 2 with the exceptions listed in the Guideline / Regulation Deviations section of this document.

M. E. Targia, B.A.

Study Director

Environmental Toxicology Laboratory

130193

DATE

PERSONNEL

Study Director

M. E. Targia, B.A.

Laboratory Head

M. L. Hinman, Ph.D.

Laboratory Supervisor

R. W. Woods, B.S.

Technicians

E. J. Febbo, B.S.

D. W. Hart, B.S.

Cs. Krisk

N. M. Roden, B.A.

E. C. Swithers, B.S.

J. Yarusinsky

Analytical and Fate Chemistry Supervisor

L. D. Twitty, A.S.

Compound Preparation Supervisor

M. A. Elliott, B.S.

Quality Assurance Supervisor

J. R. Jackson, B.S.

QUALITY ASSURANCE STATEMENT

STUDY NUMBER: 144442

TEST SUBSTANCE/ARTICLE: MRD-92-444

STUDY SPONSOR: Cytec Industries

Listed below are the dates that this study was inspected by the Quality Assurance Unit of Exxon Biomedical Sciences, Inc., and the dates findings were reported to the Study Director and Management.

Date(s) of Inspection	Reported to Study Director	Reported to Management
16-Mar-93	16-Mar-93	22, 23-Mar-93
19, 26-Mar-93	26-Mar-93	30-Mar-93
14-Apr-93	15-Apr-93	19, 21-Apr-93
16-Apr-93	19-Apr-93	23-Apr-93
17-May-93	17-May-93	01, 07-Jun-93

Joanne R. Jackson, B.S.
Quality Assurance Supervisor

17 Jun 93 Date

SUMMARY

An acute immobilization test was performed to evaluate the toxicity of the test material MRD-92-444 (CT-519-92O) to *Daphnia magna*. Due to an excessive number of control organisms (>10%) trapped at the surface and/or immobilized in the first four trials, a fifth trial was performed. The results of the fifth test are presented.

Preliminary methods development indicated that mixing the test material in dilution water for 1 hour was most appropriate for this study.

A stock solution was prepared by adding the appropriate amount of neat test material to dilution water in a glass aspirator bottle. The stock solution was mixed (<10% vortex) on a magnetic stirplate with a Teflon® coated stirbar for 1 hour. The nominal treatment levels for the test were 5mg/L, 1mg/L, 0.2mg/L, 0.04mg/L, 0.008mg/L, 0.0016mg/L and a laboratory dilution water control (BW2). Test treatments were prepared by adding the appropriate amount of stock solution to laboratory dilution water. Samples were removed from each treatment and analyzed for carbon content. The daphnids were exposed to each treatment level for a 48-hour period.

Due to the low percentage of carbon in this material (20.86%) and the variability of the analytical method at the loading levels tested, measured concentrations could not be determined. As such (and since the test material is soluble), nominal concentrations were used for the statistical analysis.

The 48-hour EC50 was calculated to be 0.6mg/L with 99% confidence intervals of 0.2 to 1.0mg/L based on the nominal concentration of the test material. The Effect Concentration (EC50) is the calculated concentration of test material that causes 50% immobilization in a population of test organisms within a specified exposure period.

INTRODUCTION

This study was conducted for Cytec Industries, 5 Garret Mountain Plaza, West Paterson, NJ 07424 to evaluate the acute toxicity of the test material MRD-92-444 (CT-519-92O) to the daphnid, Daphnia magna.

This test was conducted in general agreement with OECD¹ guidelines, and was performed to comply with OECD GLP regulations².

The study was performed by the Environmental Toxicology Laboratory of Exxon Biomedical Sciences, Inc., Mettlers Road, CN 2350, East Millstone, NJ 08875-2350. The Environmental Toxicology Laboratory is certified by the New Jersey Department of Environmental Protection and Energy for Acute Bioassay Testing.

¹Daphnia sp. Acute Immobilization Test. OECD Guidelines for Testing of Chemicals. Section 2: Effects on Biotic Systems, Guideline 202, adopted 4-Apr-84.

²OECD Principles of Good Laboratory Practice, C(81)30 (Final), Annex 2.

MATERIALS AND METHODS

Study Initiation Date

25-Feb-93

In-life Test Period

16-Apr-93 to 18-Apr-93

Experimental Termination

18-Apr-93

Test Material Identification

MRD-92-444 (CT-519-92O)

Description

Amber liquid

Storage Conditions

Room temperature

Vehicle

None

Justification of Dosing Route

Potential environmental exposure is by the test material in water.

Carrier / Dilution Water

Laboratory Dilution Water, Blend Water 2 (BW2). The dilution water is aerated by circulation within the holding tank and aged ≥24 hours before introduction to the test system.

Characterization of Test Material

The identity (including batch number and composition, purity and concentrations, where appropriate, or other characterizations to appropriately identify each batch of the test substance) and the stability are the responsibility of the Spansor

Analysis of Mixtures

Samples were removed from each treatment on Day 0 and 2 and analyzed for carbon content.

The results of these analyses are included in Appendix A on page 18.

Mixture sample retention: None

Test System

Daphnia magna

Method of Species Verification

Keyed using Fresh-Water Biology, Ward, H. B. and G. C. Whipple, 1959, John Wiley and Sons, Inc.

Justification for Selection of Test System

Daphnia magna has been used in safety evaluation and is a common test species for freshwater toxicity studies.

Supplier

Cultured in the Environmental Toxicology Laboratory of Exxon Biomedical Sciences, Inc.

Husbandry and Acclimation

Daphnids are kept in 1 liter glass culture vessels with dilution water (BW2 at 20 \pm 2°C) similar in characteristics to that used during the test.

Cultures of Daphnia are fed $\sim 3.12 \times 10^5 cells/mL$ per day of Selenastrum capricomutum and $\sim 6\mu L/mL$ of a yeast / salmon starter / cereal leaves mixture per day. Algae are cultured in the Environmental Toxicology Laboratory of Exxon Biomedical Sciences, Inc. using modified ASM-1 medium.

Number and Sex

Number: 140; Sex: Not Applicable

Age at Initiation of Exposure

<24 hours old, taken from 13-day old parents.

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Test System Identification

Test organisms were not individually identified. All test chambers were labeled to show study number, concentration, randomization number and replicate chamber number.

Selection

Organisms were randomly distributed one at a time until each chamber contained five.

Feed

Test organisms were not fed during the study.

Contaminants

The quality of the dilution water used in culture and testing is monitored at weekly, monthly, semi-annual and annual intervals (Appendix A). There are no known contaminants in the water or the feed believed to be at levels high enough to interfere with this study.

Range Finding Test

A range finding test was performed to determine the concentrations for the definitive test. Nominal concentrations were: 1.0g/L, 0.1g/L, 0.05g/L, 0.01g/L, 0.005g/L and a control of BW2. Individual treatment solutions were prepared by adding the neat test material to dilution water. These solutions were mixed for 1 hour. The WAF of these treatment mixtures was used to prepare two replicates containing five organisms. Total immobilization was observed in the 1.0g/L treatment during at 24 hours of exposure. Total immobilization was observed at all other treatment levels (except the control) at 48 hours.

NICHATATAT

Definitive Test Design

GROUP	NOMINAL CONCENTRATION (MRD-92-444) (CT-519-92O) (mg/L)	NUMBER OF TEST ORGANISMS
(CONTROL)	0	20 (5 per 4 replicates)
2	0.0016	20
3	0.008	20
4	0.04	20
. 5	0.2	20
6	1.0	20
7	5.0	20

Preparation and Administration of Test Material

A stock solution was prepared by adding the appropriate amount of neat test material to dilution water in a glass aspirator bottle. The stock solution was mixed (<10% vortex) on a magnetic stirplate with a Teflon® coated stirbar for 1 hour. The stock solution appeared clear. The stock solution was drawn through the outlet at the bottom of the vessel and added to dilution water to prepare the nominal treatment levels for the test. Treatments were then divided into 4 replicate chambers. Test chambers were covered with glass to minimize evaporation and/or volatilization.

Test Chamber / Volume

100mL glass beaker / 50mL

Exposure Duration

48 hours (\pm 1 hour)

Exposure Conditions

Mean test temperature: 20.1 ± 0.1 °C (s.d.), continuously monitored.

Diurnal light: Dark throughout the study.

Experimental Evaluation

Observations for immobilization, abnormal behavior and appearance of the daphnids were performed on all replicate chambers at 24 and 48 hours (\pm 1 hr). Immobilization is the lack of swimming ability within ~15 seconds after gentle agitation of the test chambers. No evidence of test material insolubility was observed in the test chambers. Water quality measurements (pH, dissolved oxygen and temperature) were performed on Day 0 and at termination in each treatment.

After the 48-hour period, monitoring of environmental conditions was discontinued and the test organisms were discarded.

Loading During the Definitive Study

≤1 daphnid / 2mL of solution

Disposal

Test solutions are disposed of under the supervision of the Site Hazardous Waste Coordinator of Exxon Biomedical Sciences, Inc.

RESULTS

Due to the low percentage of carbon in this material (20.86%) and the variability of the analytical method at the loading levels tested, measured concentrations could not be determined. As such (and since the test material is soluble), nominal concentrations were used for the statistical analysis.

The 24-hour Effect Concentration (EC50) value with 95% confidence intervals was determined using the Probit procedure³ of SAS⁴ and was based upon the natural log of the of the concentration. The 48-hour EC50 value with 99% confidence intervals was determined using the Binomial Method⁵. The EC50 values are as follows:

Exposure Period (hours)	<u>EC50</u> (mg/L)	Confidence Intervals (mg/L)
24	0.8	0.2 - 2.6
48	0.6	0.2 - 1.0

The minimum concentration causing 100% immobilization was 1.0mg/L. The maximum concentration causing no immobilization was 0.2mg/L.

Treatment	% Immo	bilization
(mg/L)	24 hours	48 hours
Control	0	0
0.0016	0	0
0.008	0	0
0.04	0	0
0.2	0	0
1.0	80	100
5.0	90	100

Table 1 presents the water quality values for the test. The in-life observations are presented in Table 2. Appendix A presents the analytical chemistry methods and results, and the dilution water analysis. The concentration-response curve is presented in Figure 1.

³Finney, D.J., 1971. Probit Analysis, Third Edition, London: Cambridge University Press.

⁴SAS User's Guide: Statistics, Version 5.18 Edition. SAS Institute, Inc., Cary, NC. 1985.

⁵Stephan, C.E., Methods for Calculating an LC50, Aquatic Toxicology and Hazard Evaluation, ASTM STP 634, F.L. Mayer and J.L. Hamelink, Eds., American Society for Testing and Materials, 1977, pp. 65-84.

GUIDELINE / REGULATION DEVIATIONS

It is unknown if the analysis to support the characterization of the test material was performed in a GLP compliant manner.

It cannot be confirmed that the percent carbon analysis was performed in a GLP compliant manner.

PROTOCOL DEVIATIONS

The protocol required the chambers to be 30mL beakers containing 25mL of test solution. For this study, 100mL beakers containing 50mL of test solution were used. This is believed to have had no impact on the integrity of the study.

RECORDS

All appropriate materials, methods and experimental measurements required in the protocol were recorded and documented in the raw data. Any changes, additions or revisions to the protocol were approved by the Study Director and the Sponsor Representative. These changes were documented in writing, and include the date, the signatures of the Study Director and the Sponsor Representative and the justification for the change.

A copy of the protocol, final report, raw data, computer generated listings of raw data and supporting documentation were deposited in the Archives of Exxon Biomedical Sciences, Inc.

Table 1 - Water Quality Values

Test Day: 0

Date: 16-Apr-93

Time: 1200 hrs.

5.0	8.4	7.2	20.1
1.0	8.4	7.2	20.1
7.0	8.4	7.2	20.1
0.04	8.4	7.2	20.1
0.008	8.4	7.2	20.1
0,0016	8.4	7.2	20.1
Control	9.8	7.2	20.1
Concentration (mg/L)	Dissolved Oxygen (mg/L)	Hd	Temperature (°C)

Test Day: 2

Date: 18-Apr-93

Time: 1245 hrs.

Concentration (mg/L)	Control	0.0016	0,008	0,04	0.2	1.0	5.0
Dissolved Oxygen (mg/L)	9.1	9.1	9.2	9.1	9.1	9.1	9.1
Hd	7.1	7.2	7.3	7.3	7.4	7.4	7.4
Temperature (°C)	20.0	19.5	19.5	19.5	19.5	19.5	19.5

Table 2 - In-Life Observations

Test Day: 1

Date: 17-Apr-93

Time: 1245 hrs.

Concentration (mg/L)		ပိ	Control			0.0	0.0016			0.008	80			Ö	0.04	
Replicate	1	2	3	4	1	2	3	4	-	7	3	+	-	2	3	4
Immobilization	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
Cumulative Immobilization	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
Trapped at Surface	0	0	0	0	0	O.	0	0	0	0	0	0	0	0	_	0
Normal	5	5	5	5	2	5	5	5	5	5	5	5	5	5	4	5
Survival	5	5	5	5	S	5	5	5	5	5	5	5	5	5	5	5

Concentration (mg/L)		0	0.2			-7	0.1			~·	5.0	
Replicate	-	2	3	4	1	2	3	4	1	2	3	þ
Immobilization	0	0	0	0	3	5	4	4	4	4	5	5
Cumulative Immobilization	0	0	0	0	3	5	4	4	4	4	5	5
Lethargic	0	0	0	0	0	0	0	0		-	0	0
Normal	5	5	5	5	2	0	-	-	0	0	0	0
Survival	5	5	5	5	2	0	-	_		-	0	0

Table 2 - In-Life Observations

Test Day: 2

Date: 18-Apr-93

Time: 1220 hrs.

Concentration (mg/L)		Con	Control			0.0	0.0016			0.008	80			Ö	0.04	
Replicate	Ţ	2	3	4	1	2	3	4	1	7	3	Ţ	1	2	က	7
Immobilization	0	0	0	0	0	0	0	0	0	. 0	0	0	0	0	0	0
Cumulative Immobilization	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
Normal	5	5	5	. 5	5	5	5	5	5	5	5	5	5	5	5	5
Survival	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5	5

Concentration (mg/L)		0	0.2				1.0			•	5.0	
Replicate	-	2	3	4	1	2	3	4	1	2	3	4
Immobilization	0	0	0	0	2	0	1	1	1	1	0	0
Cumulative Immobilization	0	0	0	0	5	5	5	5	5	5	5	5
Normal	5	5	5	5	0	•	0	0	0	0	•	•
Survival	5	5	5	5	0	ı	0	0	0	0	•	,

⁽⁻⁾ No observations were performed due to total mortality in replicate.

Appendix A Analytical Results

Analytical Chemistry Results

Due to the complex nature of the test material, samples of MRD-92-444 (CT-519-920) in water were analyzed for Dissolved Organic Carbon⁶ (DOC) content. DOC results were obtained by filtering the samples through a $0.45\mu m$ filter and analyzing for Total Carbon (TC) and Inorganic Carbon (IC) with the difference between the two values considered DOC. Samples were analyzed using a Dohrmann DC-190 Total Organic Carbon Analyzer.

Nominal Chemical	DOC	(ppm)
Conc. (mg/L)	Day 0	Day 2
Control	2.446 ± 0.098	2.436 ± 0.188
0.0016	2.176 ± 0.117	2.255 ± 0.194
0.008	2.010 ± 0.159	1.985 ± 0.071
0.04	2.161 ± 0.196	1.845 ± 0.117
0.2	1.783 ± 0.157	1.960 ± 0.026
1.0	1.876 ± 0.071	2.142 ± 0.193
5.0	2.004 ± 0.165	2.293 ± 0.119

⁶American Public Health Association, American Water Works Association and Water Pollution Control Federation. 1989. Standard Methods for the Examination of Water and Wastewater, 17th ed. American Public Health Association, Washington, D.C. Method 5310B, Combustion-Infrared.

Dilution Water (Carrier Water) Analysis

The dilution water used by the Environmental Toxicology Laboratory is ground water from a well located at the Environmental Toxicology Laboratory in East Millstone, NJ. The well water is treated by the system depicted in Figure A-1. The water system is composed of glass and 316 stainless steel, Teflon® and contains no materials known to leach into the water. The water used during this study was a blend of carbon filtered well water and reverse osmosis dialyzed well water.

The following water quality data is most representative of the dilution water used during the in-life period of the study. Table A-1 presents analyses performed on the blended dilution water (BW2) prior to study start. Table A-2 presents the analyses of the chemical pollutant parameters of the carbon treated well water ("SV-5"), since analyses are not performed on individual blends. Water quality analyses (dissolved oxygen, pH, alkalinity, hardness and specific conductance) are performed by the Environmental Toxicology Laboratory personnel. Monthly Total Organic Carbon analysis is performed by the Analytical Chemistry Laboratory at Exxon Biomedical Sciences, Inc. All other analyses are performed by a contracted laboratory.

Table A-1
Results of Water Quality Analysis

Sample	Sample Date	Alkalinity as CaCO ₃ (mg/L) ♦	Hardness as CaCO ₃ (mg/L) ©	Specific Conductance (µmhos)	pН	Dissolved Oxygen (mg/L)
BW2	16-Apr-93	30	100	200	7.7	8.5
BW2	22-Apr-93	31	82	200	7.6	8.7

Monthly Water Quality Analysis

Sample	Sample Date	Total Organic Carbon (ppm)*
BW2	18-Mar-93	1.142
BW2	21-Apr-93	1.660

- ◆ U.S. Environmental Protection Agency. 1979, Revised March 1983. Methods for Chemical Analysis of Water and Wastes, EPA-600/4-79-020. Office of Research and Development, Cincinnati, OH. Method 310.1, Alkalinity (Titrametric, pH 4.5).
- □ U.S. Environmental Protection Agency. 1979, Revised March 1983. Methods for Chemical Analysis of Water and Wastes, EPA-600/4-79-020. Office of Research and Development, Cincinnati, OH. Method 130.2, Alkalinity (Titrametric, EDTA).
- * American Public Health Association, American Water Works Association and Water Pollution Control Federation. 1989. Standard Methods for the Examination of Water and Wastewater, 17th ed. American Public Health Association, Washington, D.C. Method 5310B, Combustion-Infrared.

Table A-2 Priority Pollutants

Semi-annual Dilution Water Analysis Base/Neutral Compounds

Description	Unit	MDL Sa	mpled 27-JAN-93 Well Water
Acenaphthene	$\mu { m g}/{ m L}$	1.9	ND
Acenaphthylene	$\mu g/L$	3.5	ND
Anthracene	$\mu g/L$	1.9	ND
Benzidine	$\mu g/L$	44.	ND
Benzo(a)anthracene	$\mu g/L$	7.9	ND
Benzo(a)pyrene	$\mu g/L$	2.5	ND
Benzo(b)fluoranthene	$\mu g/L$	4.8	ND
Benzo(ghi)perylene	$\mu g/L$	4.1	ND
Benzo(k)fluoranthene	$\mu g/L$	2.5	ND
bis(2-Chloroethoxy)methane	$\mu g/L$	5.4	ND
bis(2-Chloroethyl)ether	$\mu g/L$	5.8	ND
bis(2-Chloroisopropyl)ether	$\mu g/L$	5.8	ND
bis(2-Ethylhexyl)phthalate	$\mu \mathrm{g}/\mathrm{L}$	10.	ND
4-Bromophenyl phenyl ether	$\mu g/L$	1.9	ND
Butyl benzyl phthalate	$\mu g/L$	10.	ND
2-Chloronaphthalene	$\mu { m g}/{ m L}$	1.9	ND
4-Chlorophenyl phenyl ether	$\mu { m g}/{ m L}$	4.2	ND
Chrysene	$\mu { m g}/{ m L}$	2.5	ND
Dibenzo(a,h)anthracene	$\mu \mathrm{g}/\mathrm{L}$	2.5	ND
1,2-Dichlorobenzene	$\mu { m g}/{ m L}$	1.9	ND
1,3-Dichlorobenzene	$\mu { m g}/{ m L}$	1.9	ND
1,4-Dichlorobenzene	$\mu { m g}/{ m L}$	4.4	ND
3,3'-Dichlorobenzidine	$\mu { m g}/{ m L}$	16.7	ND
Diethyl phthalate	$\mu \mathrm{g}/\mathrm{L}$	10.	ND
Dimethyl phthalate	$\mu { m g}/{ m L}$	5.1	ND
Di-n-butyl phthalate	$\mu \mathrm{g}/\mathrm{L}$	10.	ND
2,4-Dinitrotoluene	$\mu { m g}/{ m L}$	5.8	ND
2,6-Dinitrotoluene	$\mu \mathrm{g}/\mathrm{L}$	1.9	ND
Di-n-octyl phthalate	$\mu g/L$	10.	ND
1,2-Diphenylhydrazine	$\mu \mathrm{g}/\mathrm{L}$	10.	ND
Fluoranthene	$\mu \mathrm{g}/\mathrm{L}$	2.2	ND
Fluorene	$\mu \mathrm{g}/\mathrm{L}$	1.9	ND
Hexachlorobenzene	$\mu \mathrm{g}/\mathrm{L}$	1.9	ND
Hexachlorobutadiene	$\mu { m g}/{ m L}$	0.91	ND
Hexachlorocyclopentadiene	$\mu \mathrm{g}/\mathrm{L}$	10.	ND
Hexachloroethane	$\mu g/L$	1.6	ND
Indeno(1,2,3-c,d)pyrene	$\mu g/L$	3.7	ND
Isophorone	$\mu g/L$	2.2	ND
Naphthalene	$\mu g/L$	1.6	ND

MDL = Minimum Detection Limits

ND = None Detected

Table A-2 Priority Pollutants (continued)

Description	Unit	MDL	Sampled 27-JAN-93 Well Water
Nitrobenzene	μg/L	1.9	ND
N-Nitrosodimethylamine	μg/L	10.	ND
N-Nitrosodi-n-propylamine	μg/L	10.	ND
N-Nitrosodiphenylamine Phenanthrene	μg/L μg/L μg/L	1.9 5.5	ND ND
Pyrene	μg/L	1.9	ND
1,2,4-Trichlorobenzene	μg/L	1.9	ND

Pesticides/PCB Compounds

Description	Unit	MDL S	ampled 27-JAN-93 Well Water
Aldrin	$\mu \mathrm{g}/\mathrm{L}$	0.051	ND
Alpha-BHC	$\mu g/L$	0.051	ND
Beta-BHC	$\mu g/L$	0.051	ND
Gamma-BHC	$\mu g/L$	0.051	ND
Delta-BHC	$\mu g/L$	0.051	ND
Chlordane	$\mu g/L$	1.0	ND
4,4'-DDT	$\mu g/L$	0.10	ND
4,4'-DDE	$\mu { m g}/{ m L}$	0.10	ND
4,4'-DDD	$\mu { m g}/{ m L}$	0.10	ND
Dieldrin	μ g/ $ m L$	0.10	ND
Endosulfan I	$\mu \mathrm{g}/\mathrm{L}$	0.051	ND
Endosulfan II	$\mu { m g}/{ m L}$	0.10	ND
Endosulfan sulfate	$\mu { m g}/{ m L}$	0.10	ND
Endrin	$\mu \mathrm{g}/\mathrm{L}$	0.10	ND
Endrin aldehyde	$\mu \mathrm{g}/\mathrm{L}$	0.10	ND
Heptachlor	$\mu \mathrm{g}/\mathrm{L}$	0.051	ND
Heptachlor epoxide	$\mu \mathrm{g}/\mathrm{L}$	0.051	ND
Aroclor-1242	$\mu \mathrm{g}/\mathrm{L}$	0.51	ND
Aroclor-1254	$\mu \mathrm{g}/\mathrm{L}$	1.0	ND
Aroclor-1221	$\mu { m g}/{ m L}$	0.51	ND
Aroclor-1232	$\mu { m g}/{ m L}$	0.51	ND
Aroclor-1248	$\mu { m g}/{ m L}$	0.51	ND
Aroclor-1260	$\mu \mathrm{g}/\mathrm{L}$	1.0	ND
Aroclor-1016	$\mu \mathrm{g}/\mathrm{L}$	0.51	ND
Toxaphene	$\mu \mathrm{g}/\mathrm{L}$	2.0	ND
Endrin ketone	$\mu {\sf g}/{\sf L}$	0.10	ND
Methoxychlor	$\mu \mathrm{g}/\mathrm{L}$	0.51	ND

MDL = Minimum Detection Limits

ND = None Detected

Table A-2 Priority Pollutants (continued)
Acid Compounds

Description	Unit	MDL	Sampled 27-JAN-93 Well Water
2-Chlorophenol	$\mu \mathrm{g}/\mathrm{L}$	3.3	ND
2,4-Dichlorophenol	$\mu \mathrm{g}/\mathrm{L}$	2.7	ND
2,4-Dimethylphenol	$\mu {\sf g}/{\sf L}$	2.7	ND
4,6-Dinitro-o-cresol	$\mu g/L$	24.	ND
2,4-Dinitrophenol	$\mu {\sf g}/{\sf L}$	42.	ND
2-Nitrophenol	$\mu {\sf g}/{\sf L}$	3.6	ND
4-Nitrophenol	$\mu \mathrm{g}/\mathrm{L}$	2.4	ND
p-Chloro-m-cresol	$\mu { m g}/{ m L}$	3.0	ND
Pentachlorophenol	$\mu \mathrm{g}/\mathrm{L}$	3.6	ND
Phenol	$\mu \mathrm{g}/\mathrm{L}$	1.5	ND
2,4,6-Trichlorophenol	$\mu g/L$	2.7	ND

Volatile Compounds

Description	Unit	MDL	Sampled 27-JAN-93 Well Water
Acrolein	$\mu \mathrm{g}/\mathrm{L}$	100.	ND
Acrylonitrile	$\mu g/L$	100.	ND
Benzene	$\mu g/L$	4.4	ND
bis(Chloromethyl)ether	$\mu g/L$	10.	ND
Bromoform	$\mu g/L$	4.7	ND
Carbon tetrachloride	$\mu g/L$	2.8	ND
Chlorobenzene	$\mu g/L$	6.0	ND
Chlorodibromomethane	$\mu g/L$	3.1	ND
Chloroethane	$\mu g/L$	10.	ND
2-Chloroethylvinyl ether	$\mu \mathrm{g}/\mathrm{L}$	10.	ND
Chloroform	$\mu \mathrm{g}/\mathrm{L}$	1.6	ND
Dichlorobromomethane	$\mu \mathrm{g}/\mathrm{L}$	2.2	ND
Dichlorodifluoromethane	$\mu \mathrm{g}/\mathrm{L}$	10.	ND
1,1-Dichloroethane	$\mu { m g}/{ m L}$	4.7	ND
1,2-Dichloroethane	$\mu g/L$	2.8	ND
1,1-Dichloroethylene	$\mu { m g}/{ m L}$	2.8	ND
1,2-Dichloropropane	$\mu { m g}/{ m L}$	6.0	ND
cis-1,3-Dichloropropylene	$\mu \mathrm{g}/\mathrm{L}$	5.0	ND
Ethylbenzene	$\mu { m g}/{ m L}$	7.2	ND
Methyl bromide	$\mu {\sf g}/{\sf L}$	10.	ND
Methyl chloride	$\mu {\sf g}/{\sf L}$	10.	ND
Methylene chloride	$\mu { m g}/{ m L}$	2.8	7.99

MDL = Minimum Detection Limits

ND = None Detected

Table A-2 Priority Pollutants (continued)

1,1,2,2-Tetrachloroethane $\mu g/L$ 6.9NDTetrachloroethylene $\mu g/L$ 4.1NDToluene $\mu g/L$ 6.0ND1,2-Trans-dichloroethylene $\mu g/L$ 1.6ND1,1,1-Trichloroethane $\mu g/L$ 3.8ND1,1,2-Trichloroethane $\mu g/L$ 5.0NDTrichloroethylene $\mu g/L$ 1.9NDTrichlorofluoromethane $\mu g/L$ 10.NDVinyl chloride $\mu g/L$ 10.NDtrans-1,3-Dichloropropylene $\mu g/L$ 10.ND	Description	Unit	MDL	Sampled 27-JAN-93 Well Water
	Tetrachloroethylene Toluene 1,2-Trans-dichloroethylene 1,1,1-Trichloroethane 1,1,2-Trichloroethane Trichloroethylene Trichlorofluoromethane	μg/L μg/L μg/L μg/L μg/L μg/L μg/L	4.1 6.0 1.6 3.8 5.0 1.9	ND ND ND ND ND ND

Metals, Cyanides, Phenols

Description	Unit	MDL	Sampled 27-JAN-93 Well Water
Antimony	μg/L	60.	ND
Arsenic	$\mu g/L$	10.	BMDL
Beryllium	$\mu g/L$	1.0	ND
Cadmium	$\mu g/L$	2.0	ND
Chromium	$\mu g/L$	10.	ND
Copper	$\mu \mathrm{g}/\mathrm{L}$	10.	ND
Lead	$\mu \mathrm{g}/\mathrm{L}$	5.0	ND
Mercury	$\mu { m g}/{ m L}$	0.20	BMDL
Nickel	$\mu { m g}/{ m L}$	20.	ND
Selenium	$\mu { m g}/{ m L}$	10.0	ND
Silver	$\mu { m g}/{ m L}$	10.	ND
Thallium	$\mu { m g}/{ m L}$	10.	ND
Zinc	$\mu { m g}/{ m L}$	20.	ND
Cyanide, Total	mg/L	0.025	<.025
Phenolics, Total	mg/L	0.050	<.050

Pesticides

Description	Unit	MDL	Sampled 27-JAN-93 Well Water
Carbophenothion	μg/L	10.	ND
Thionazin	$\mu g/L$	1.0	ND
Dimethoate	$\mu g/L$	2.5	ND
Disulfoton	$\mu g/L$	0.51	ND
Methyl parathion	$\mu g/L$	1.0	ND
Parathion	$\mu g/L$	1.0	ND
Phorate	$\mu g/L$	2.5	ND
Famphur	$\mu g/L$	10.	ND
Tetraethylpyrophosphate	$\mu g/L$	2.5	ND
BMDL = Below Minimum Detection Limits ND = None Detected	MDL	= Minimum D	Petection Limits

Table A-2 Priority Pollutants (continued)

Herbicides

Description	Unit	MDL	Sampled 27-JAN-93 Well Water
2,4-D	μg/L	3.6	ND
2,4,5-TP (Silvex)	μg/L	0.71	ND

Miscellaneous Analyses

Unit	MDL	Sampled 27-JAN-93 Well Water
mg/L	.05	.07
% of total		0.914
mg/L	4.	<4
mg/L	0.1	< .1
	mg/L % of total mg/L	mg/L .05 % of total mg/L 4.

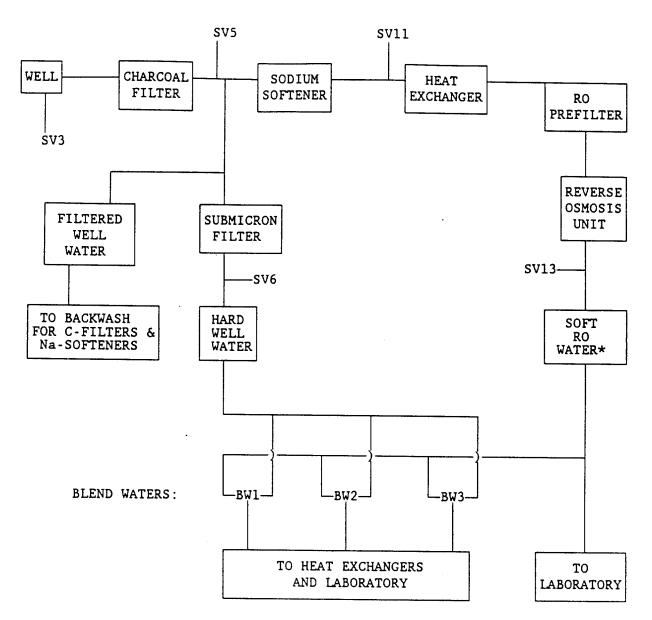
Annual Analyses

Description	Unit	MDL	Reverse Osmosis Water
Standard Plate Count ^{A1}	col/mL	1.	< 1.0
Water Suitability Test ^{A1} (Microbacterial Properties)		(Standard) 0.8-3.0	(Ratio "A") 1.19

MDL = Minimum Detection Limits
ND = None Detected

A1. performed on SV13 (Reverse Osmosis water); sampled 27-Jan-93

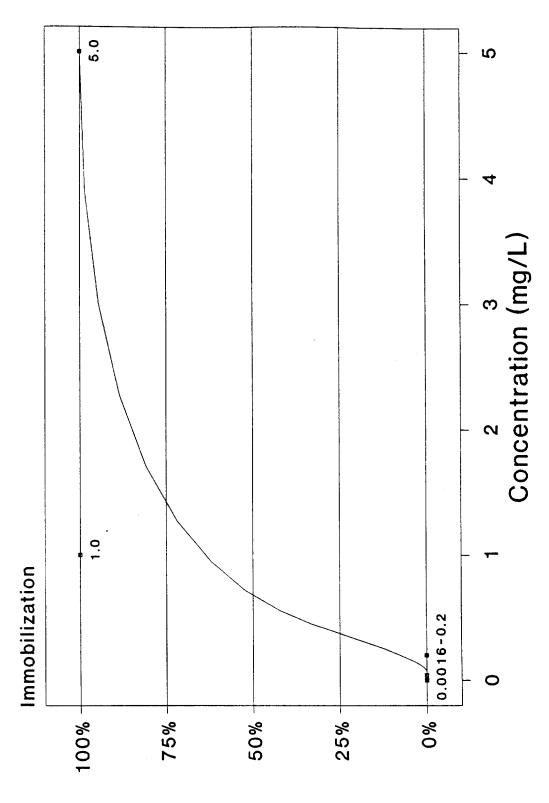
Figure A-1 Environmental Toxicology Laboratory Water System



* SOFTENED WATER OF ZERO HARDNESS

Figure 1 Concentration-Response Curve

DAPHNIA ACUTE IMMOBILIZATION TEST WITH MRD-92444 (CT-519-920)





UNITED STATES ENVIRONMENTAL PROTECTION AGENCY

WASHINGTON, D.C. 20460

OFFICE OF PESTICIDES AND TOXIC SUBSTANCES

H. Michael D. Utidjian, M.D. Corporate Medical Director American Cyanamid Company One Cyanamid Plaza Wayne, New Jersey 07470

FEB 2 2 1994

This letter formally acknowledges EPA's receipt of information submitted by your organization under Section 8(e), the "substantial risk" information reporting provision of the Toxic Substances Control Act (TSCA). For your reference, copies of the first page(s) of your submission(s) are enclosed and display the TSCA Section 8(e) Document Control Number (i.e., 8EHQ-0000-0000 Init.) assigned by EPA to your submission(s). Please refer to this cited number when submitting follow-up or supplemental information.

Please note that all submitted correspondence will be placed in the public files unless confidentiality is claimed according to the procedures outlined in Part X of EPA's TSCA Section 8(e) policy statement (43 FR 11110, March 16, 1978).

Confidential submissions submitted pursuant to the TSCA Section 8(e) Compliance Audit Program (CAP) should already contain information supporting confidentiality claims, because substantiation of CBI claims is required at the same time the 8(e) CAP is submitted to EPA. (If not done so already, please ensure that this information is provided to the Agency). When substantiating any/all claims, answer the questions detailed in the following attachment.

For NON-CLP submissions, any confidentiality claims should be supported by submission of information as described in the attachment(s).

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7 SEO A	OTS DATE: 11 04	•			P F C INFOR	01 02 04 0216 01 02 04 0217 01 02 04 0218 01 02 04 0220 01 02 04 0222 01 02 04 0222 01 02 04 0223 01 02 04 0223 01 02 04 0224 01 02 04 0228 01 02 04 0228 01 02 04 0228 01 02 04 0228 01 02 04 0228 01 02 04 0228 01 02 04 0228 01 02 04 0228 01 02 04 0228 01 02 04 0228 01 02 04 0228 01 02 04 0228 01 02 04 0228 01 02 04 0228 01 02 04 0228 01 02 04 0228 01 02 04 0228	KLITK
SUBMITTER NAME: AMERICAN COMPANY	SUB. DATE: 11 03 93	CHEMICAL NAME:		10 RD - 92 - 4444	INFORMATION TYPE:	0201 ONCO (HUMAN) 0202 ONCO (ANIMAL) 0204 CLIL TRANS (IN VITRO) 0205 MUTA (IN VITRO) 0206 REPRO/TERATO (HUMAN) 0206 REPRO/TERATO (HUMAN) 0207 REPRO/TERATO (HUMAN) 0209 NEURO (HUMAN) 0210 ACUTE TOX. (HUMAN) 0211 CHR. TOX. (HUMAN) 0212 ACUTE TOX. (ANIMAL) 0213 SUB ACUTE TOX (ANIMAL) 0214 SUB CHRONIC TOX (ANIMAL) 0215 CHRONIC TOX (ANIMAL) 0215 CHRONIC TOX (ANIMAL) 0216 NUBCRE DATA: NON-CBI INVENTORY TRIAGE DATA: NON-CBI INVENTORY	IN DAMINE

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